TixSi1-xO2 deposited by plasma-enhanced atomic layer deposition for spacerapplication

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Thin film grown by AtomicLayer Deposition has been enabled advanced nano-patterning technology such as spacerdefined multiple patterning. The main scheme of this patterning technology isusing sidewall spacer deposited by ALD as a hardmask. However, usually the high aspect ratio and poor mechanical strength of spacersoften causes its collapse problem and this phenomenon was also identified duringmultiple patterning process. Therefore, the studies for spacer materials that have high etchselectivity and good mechanical properties are required, but there is lack ofresearch on it. In this respect,  $TiO_2$  is one of good candidate forspacer materials. Here, we studied on film properties of  $TiO_2$  grown by ALD forspacer materials and further investigations

for mixture with SiO<sub>2</sub>were followed. We observed growth of Ti<sub>x</sub>Si<sub>1-x</sub>O<sub>2</sub>(x=0~1) using Ti(CpMe<sub>5</sub>)(OMe)<sub>3</sub>, Ti(O<sup>i</sup>Pr)<sub>4</sub>and H<sub>2</sub>Si[N(C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>]<sub>2</sub> andO<sub>2</sub> gas using PE-ALD at low temperature (100 °C). The chemical composition and carbon impurities of the films were analyzed by x-ray photoelectron spectroscopy (XPS), and the nanostructures of the films were analyzed by x-ray diffraction (XRD). And mechanical property ofTi<sub>x</sub>Si<sub>1-x</sub>O<sub>2</sub> films was investigated bynanoindentation. We compared the etch rate of Ti<sub>x</sub>Si<sub>1-x</sub>O<sub>2</sub> films using both dry and wet etching process. As a result, we obtained Ti<sub>x</sub>Si<sub>1-x</sub>O<sub>2</sub> films with various Ti/(Ti+Si) compositions and there was no Ti precursordependency on dry etch rate which decreases as Ti composition increases. However, both pure TiO<sub>2</sub> films were not strippable by diluted HF solution dueto its anatase phase. Furthermore, wet etch rate of mixtures were higher than evenpure SiO <sub>2</sub> film when the films deposited by using Ti(CpMe<sub>5</sub>)(OMe)<sub>3</sub>.