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Power-to-gas conversion is highlighted where water electrolysis driven by renewable electricity produces hydrogen. Renewable hydrogen is directly used to drive fuel cell, or combined with catalytic conversion of CO<sub>2</sub> into CH<sub>4</sub> that is distributed through existing gas grid for widespread use. Based on the similar concept, we propose CH<sub>4</sub> conversion with CO<sub>2</sub> using nonthermal plasma enhanced catalytic reaction. This technology is known as dry methane reforming (DMR) and a mixture of CO and H<sub>2</sub> (syngas) is synthesized. Analogous to electrolysis of water splitting, renewable electricity is converted into chemical energy via nonthermal plasma assisted endothermic reaction. Syngas is converted preferably into carbon containing liquid fuels such as gasoline and methanol whose energy density is 10-100 times greater than that of secondary batteries. Transport and storage capability of renewable electricity is improved greatly and CO<sub>2</sub> utilization is realized simultaneously. Currently, the electrochemical reaction is dominantly studied for electrical-to-chemical energy conversion. Besides, nonthermal plasma provides additional energy and material conversion pathways, contributing to an extended carbon recycling network and fuel flexibility via the electrification of thermochemical reaction system. In this presentation, pulsed dry methane reforming (DMR; CH<sub>4</sub>+ CO<sub>2</sub> = 2CO + 2H<sub>2</sub>) in Ni/Al<sub>2</sub>O<sub>3</sub> catalysts and dielectric barrier discharge (DBD) hybrid reaction is presented with deep insight into reaction mechanisms.